# **S5P1: Matter Study Guide**

There are 3 states of matter:

Has a definite	Has a definite	Has no definite	
and	but no definite	or The	
The particles fit very closely	The	particles are the farthest	
together and stay put.	particles are close together,	apart and move the	
	but can move past each		
	other.		
Ice to become water, then can		to become a gas. Water	
vapor can turn to liquid through		An example of this would	
be on the grass, drops of water forming on a glass, or water drople		glass, or water droplets	
forming on the mirror or wall after a shower. Water below 32°F.			
Mixtures & Solutions:			
A mixture is made when two or more substances are combined, but they can be			

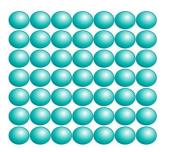
A solution is a mixture where one of the substances \_\_\_\_\_\_ in the other.

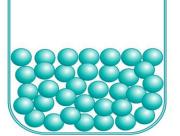
Physical Change	Chemical Change
<ul> <li>Something that changes the physical</li> </ul>	<ul> <li>A reaction that rearranges the</li> </ul>
property of a substance, such as	particles of two or more substances to
,, or	create a substance.
·	Examples include
Examples include changes in	changes, producing a,
of matter	,, or
(, melting,	·
evaporation,)	

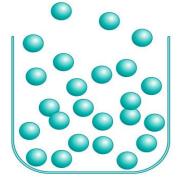
### **S5P1: Matter Study Guide**

There are 3 states of matter:

Solid	Liquid	Gas
Has a definite <b>volume</b> and	Has a definite <b>volume</b> but	Has no definite <b>volume</b> or
shape.	no definite <b>shape.</b> The	<b>shape</b> . The particles are the
The particles fit very closely	particles are close together,	farthest apart and move the
together and stay put.	but can move past each	fastest.
	other.	







Ice melts to become water, then can evaporate to become a gas. Water vapor can turn

to liquid through condensation. An example of this would be dew on the grass, drops of

water forming on a **cold** glass, or water droplets forming on the mirror or wall after a shower.

Water freezes below 32°F.

## Mixtures & Solutions:

A mixture is made when two or more substances are combined, but they can be

### separated.

A solution is a mixture where one of the substances **dissolves** in the other.

Physical Change	Chemical Change
<ul> <li>Something that changes the physical</li> </ul>	<ul> <li>A reaction that rearranges the</li> </ul>
property of a substance, such as	particles of two or more substances to
<b>shape, size</b> , or <b>form</b> .	create a <b>new</b> substance.
Examples include changes in <b>states</b>	<ul> <li>Examples include color changes,</li> </ul>
of matter ( <b>freezing</b> , melting,	producing a <b>gas, rust, heat</b> or <b>odor</b> .
evaporation, <b>condensation</b> )	